

ESCAP

Conference on Transboundary Air Pollution in North-East Asia

Common Meeting Room (#1014), METI, Tokyo, Japan

17-19 December 2008

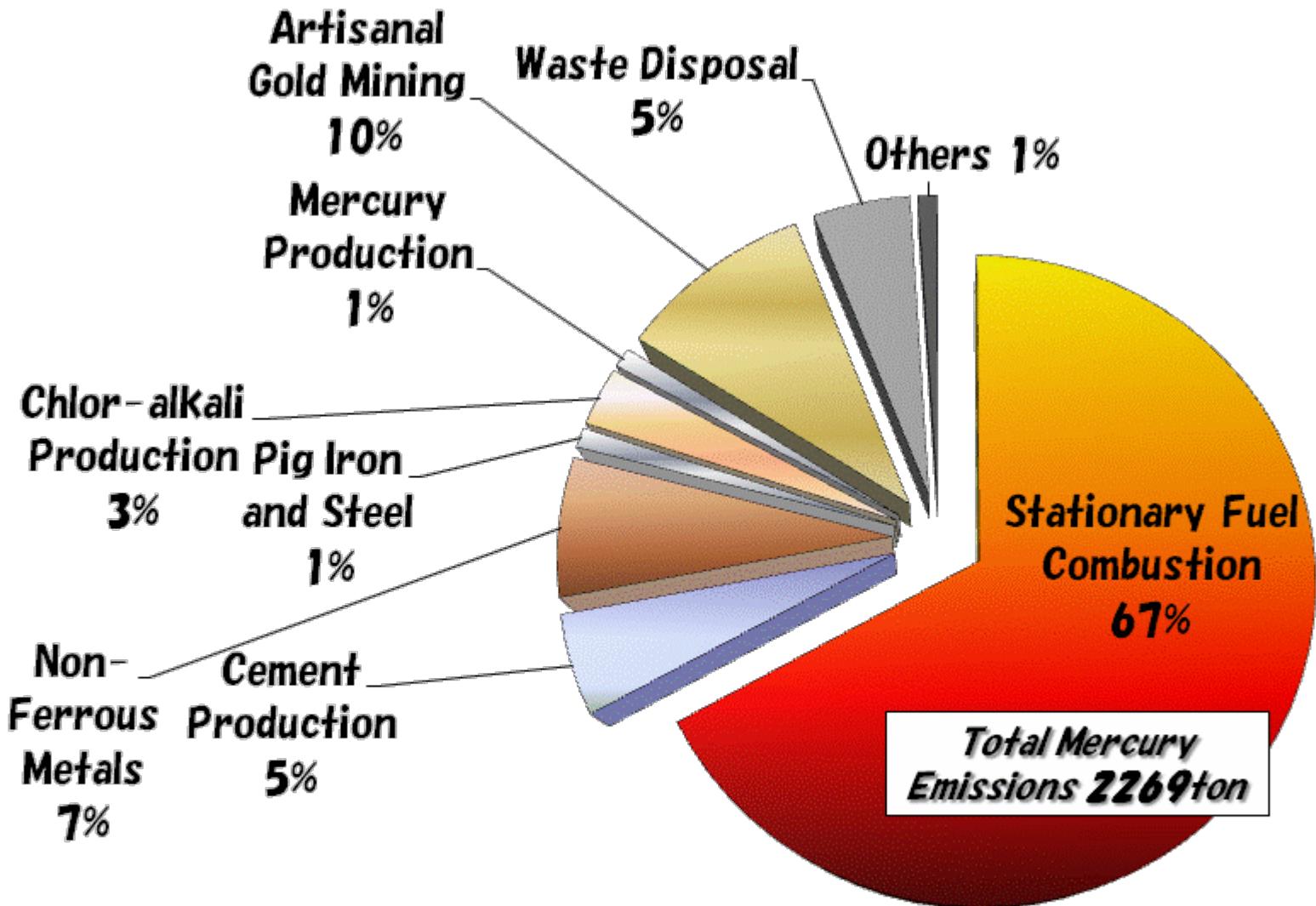
Session 1

The state of transboundary air pollution in Northeast Asia

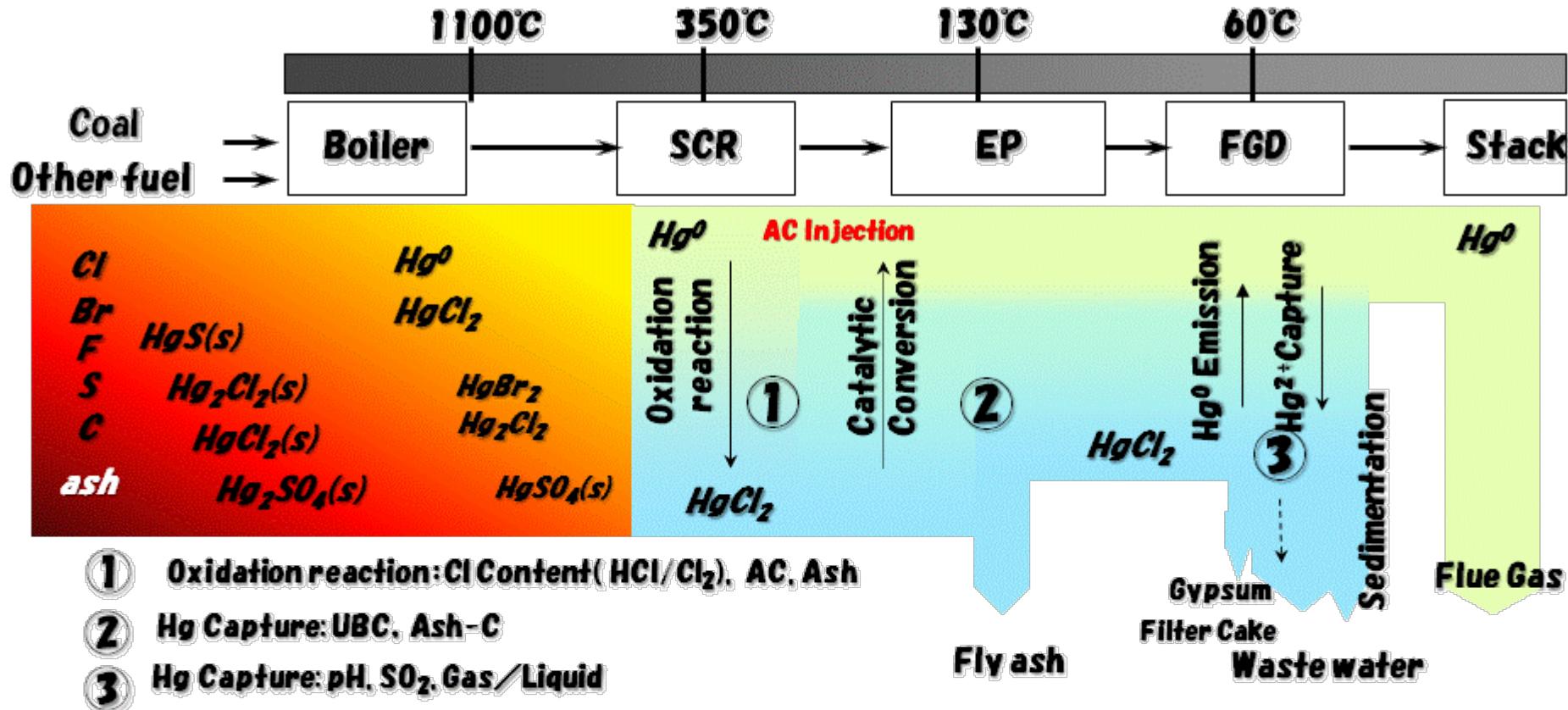
Mercury Emission from Coal Combustion in Japan

Prof. Hiroshi Moritomi

Gifu University, Graduate School of Engineering, ERES



**Fig. Anthropogenic mercury emissions,
total 2269 ton at 2000.**



Combustor type	Fuel	Coal cleaning	Bottom ash	EP	FGD	Stack	
PC	Sub-bituminous	10-50%	(10%)	0-27%	-63%	10-81%	USEPA(1998.2002)
PC	Bituminous	10-50%	(10%)	18-81%	1-41%	2-52%	USEPA(1998.2002)
Stoker			17%	17%(Dust)		56%	Wang et al.(2000)
Small PC			7%	23%		70%	Wang et al(2000)
MSW Incineration				30-60%	6-40%	15-60%	Pirrone et al.(2001)
Incineration			1.8%	13.9%	77%	7.3%	Nakamura(1994)

Fig. Effects of Flue Gas Treatment Equipment on Hg Reduction

Table Control Technology of Mercury Emission

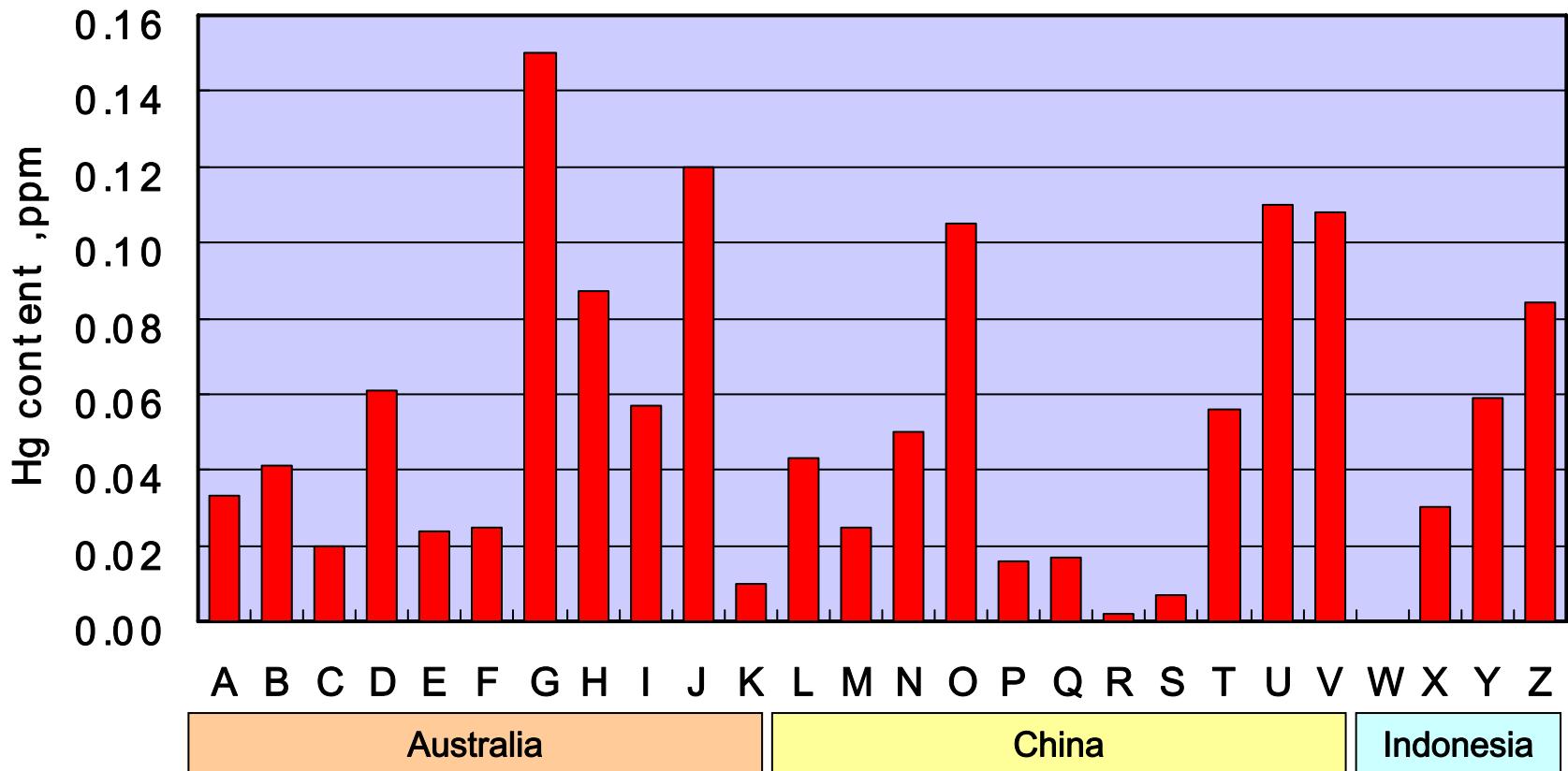
Technology	Efficiency	Other effects
DeNOx (SCR)	Unknown	30-60% NO _x
DeNOx (SCR)	SCR +Wet Scrubber	70-90% NO _x
Low NO _x burner	Unknown	>50% NO _x
Coal Cleaning	0-78%	48% SO ₂
Wet scrubber	>90% Hg ⁰ (No Hg ²⁺)	80-90% SO ₂
SCR +Wet scrubber	>80% Hg Bituminous coal	>90% SO ₂ and >90% NO _x
Dry scrubber + EP + Baghouse	6-9% USEPA 約63%	80-90% SO ₂
ESP	0-82% (Low temp ESP) USEPA 36% Bituminous USEPA 3% Sub-bituminous	>99% PM
Baghouse	0-73% USEPA 90% Bituminous USEPA 72% Sub-bituminous	>99% PM
High efficiency EP	0-50% (in testing)	>99% PM
Wet EP	約 30% (in testing)	56% PM (in testing)
EP + Baghouse	34-87% (in testing)	>99%(in testing)
AC Injection	80% Bituminous+EP+COHPAC 55-60% Sub-bituminous+EP	—
Changing fuel	>99% LNG	>99% SO ₂ and PM control; 50-75% NO _x

Mercury emission from coal power station

	USA	Japan
Mercury in coal	85ppb	50ppb(2 /3)
Mercury from Stack	5 4 %	30%(1/2)
Coal consumption	900MT	1 0 0MT(1/ 9)
Mercury emission	41t/y	1.5t/y(1/27)

USA : EPA(1999), Japan : Idemitsu
Kosan

Mercury content in coals used in Japan

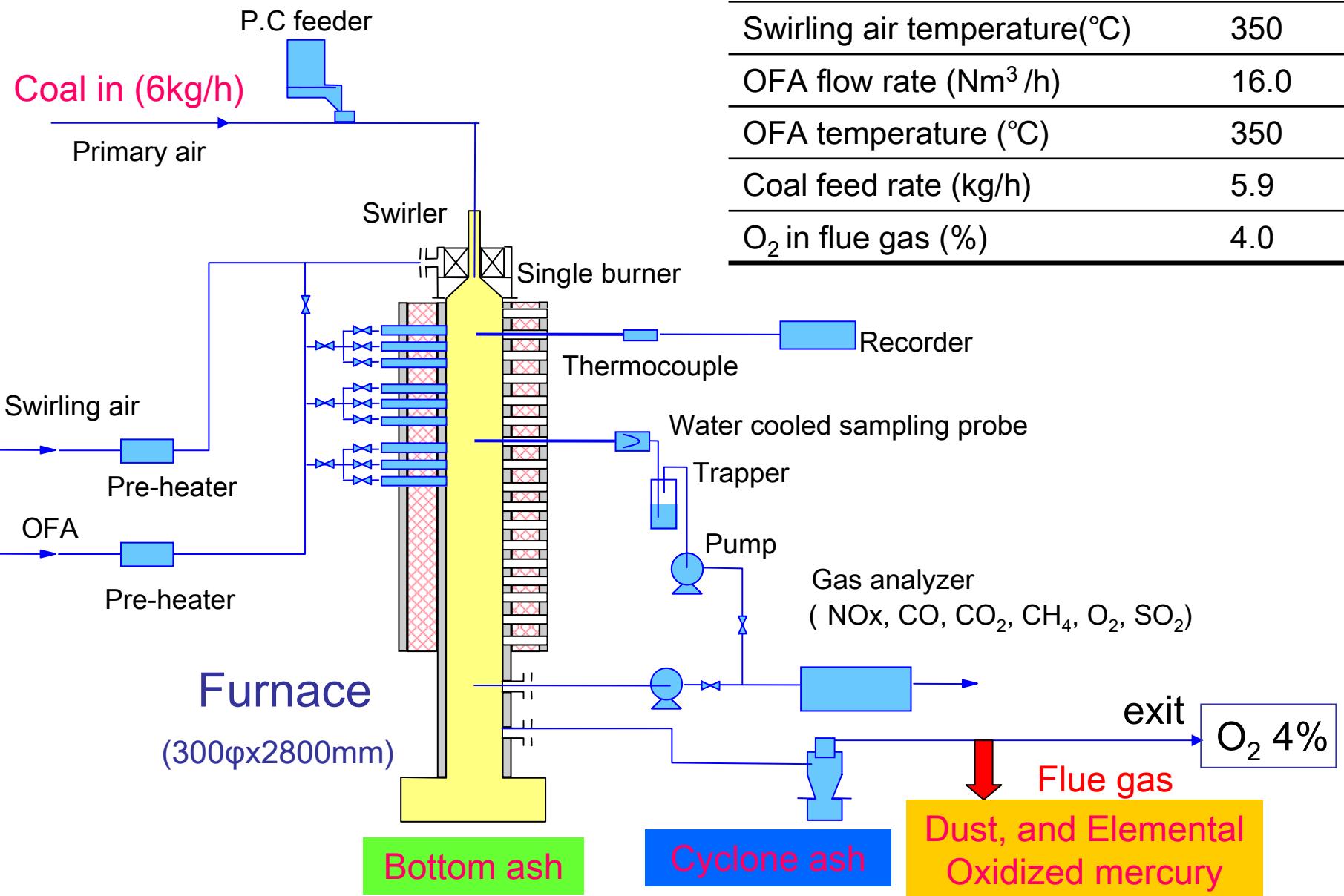


Mean content : 47 ($\mu\text{g}/\text{kg}$, ppb), Range : <10 ~ 190 ($\mu\text{g}/\text{kg}$)

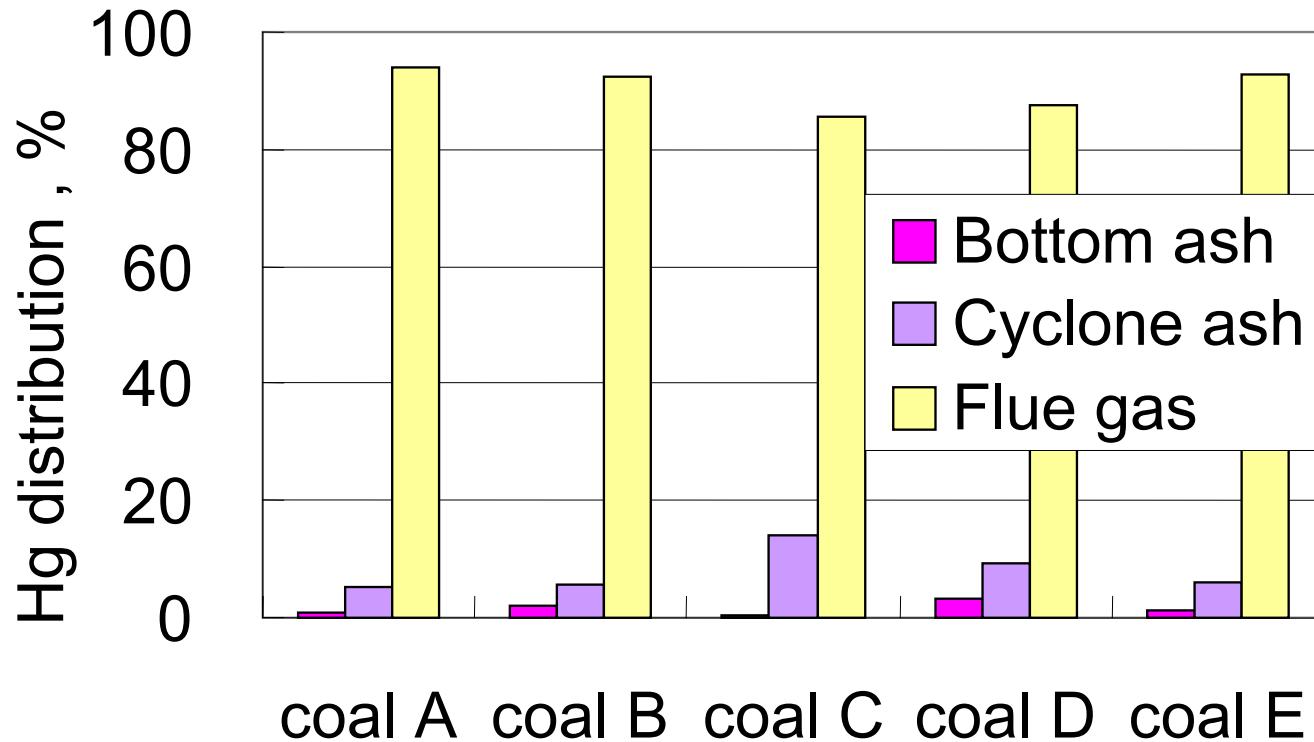
Coal properties

	Proximate analysis , as received wt%				Ultimate analysis, daf wt%					ppm	ppb
	Volatile Matter	Fixed Carbon	Moisture	Ash	C	H	O	N	S	Cl	Hg
Coal-A	33.1	55.4	1.7	9.8	82.8	5.3	9.8	1.6	0.6	233	114
Coal-B	27.4.	57.7	2.3	12.6	85.4	5.2	7.3	1.9	0.3	408	49
Coal-C	27.6	60.0	4.2	8.2	82.9	4.8	10.0	2.0	0.3	176	29
Coal-D	26.0	58.8	4.6	10.6	81.1	4.4	12.0	1.8	0.67	2304	44
Coal-E	40.9	41.5	3.1	14.5	78.2	5.9	13.6	1.3	1.08	176	119

Bench-scale PCF

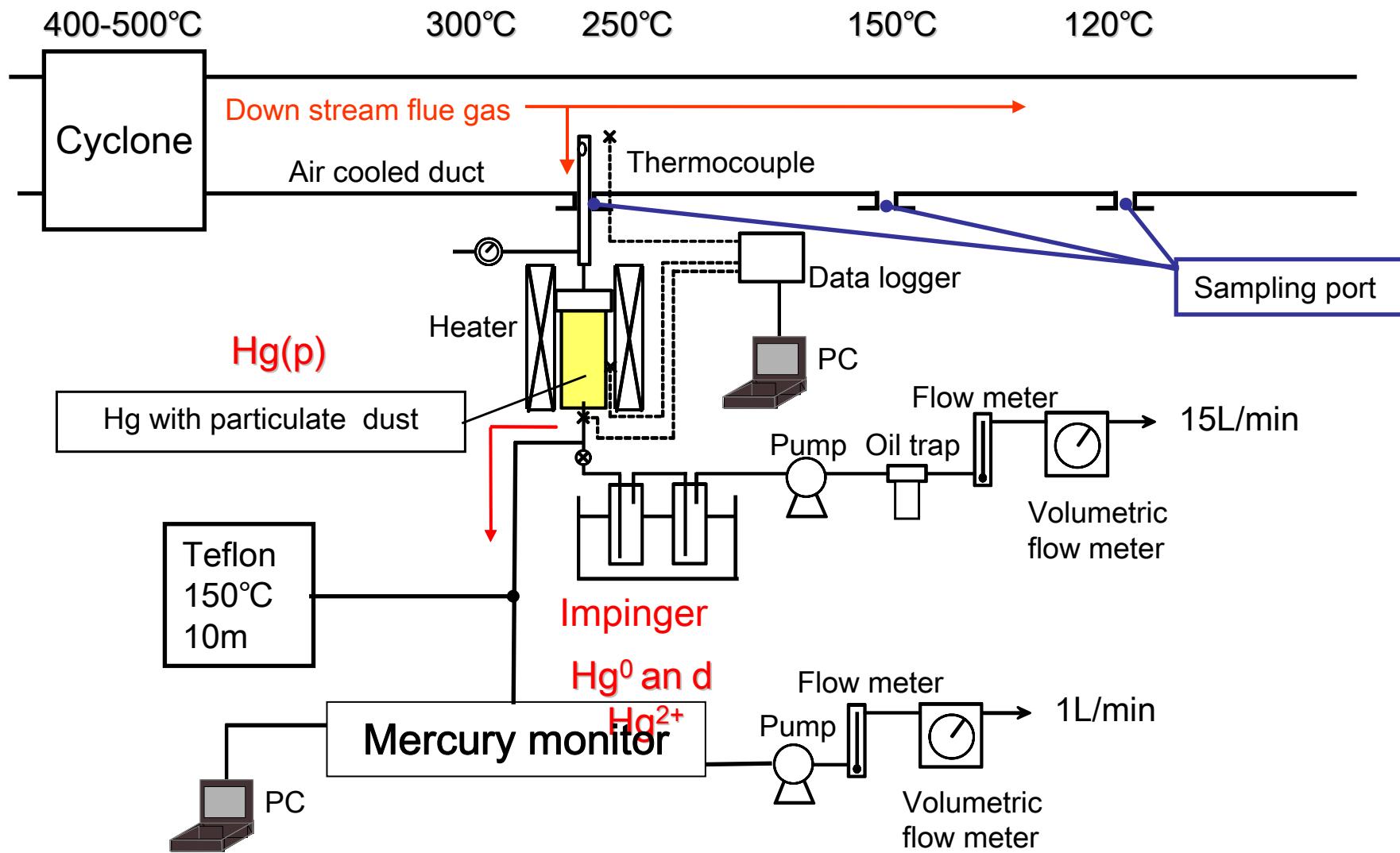


1) Mercury speciation with large particles at medium temperature

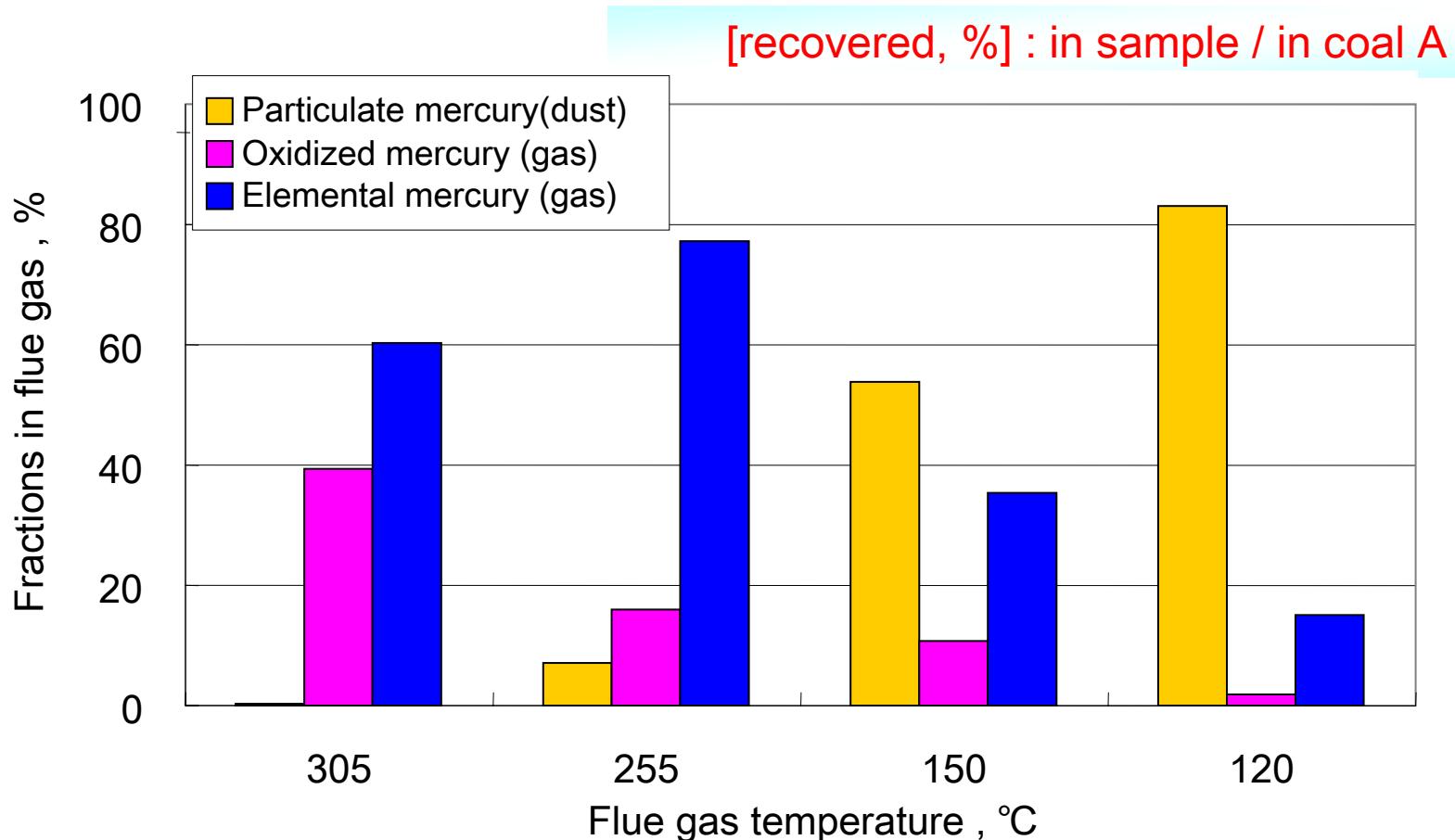


Almost of Hg to flue gas >> BA and CA

Flue gas sampling system

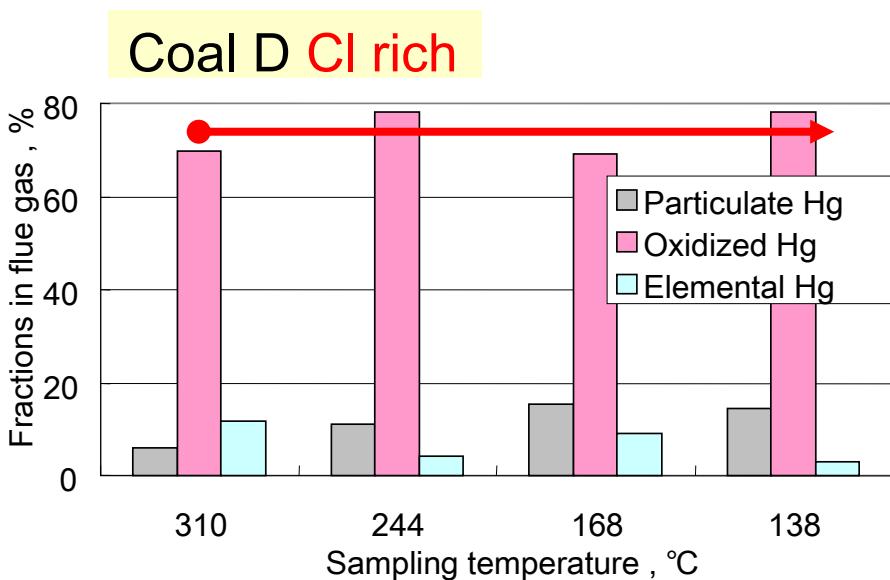
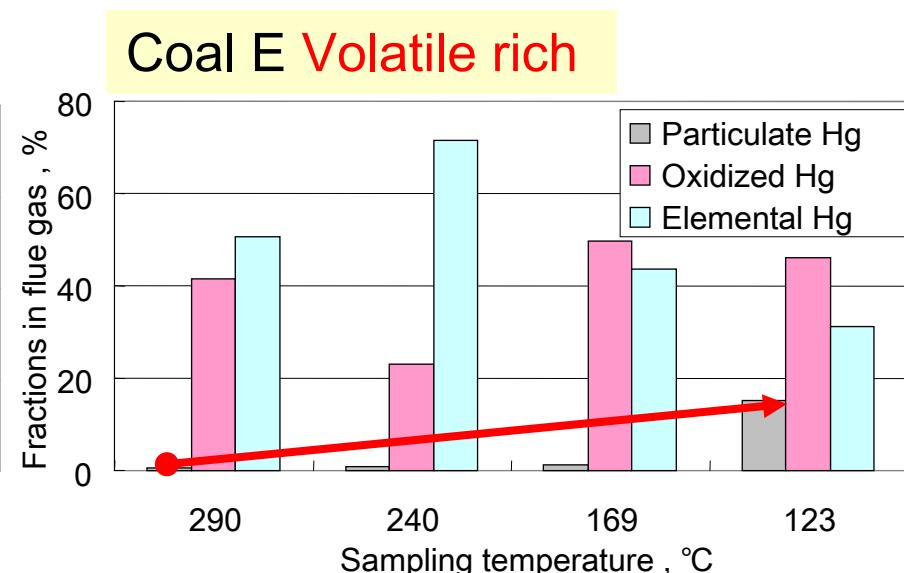
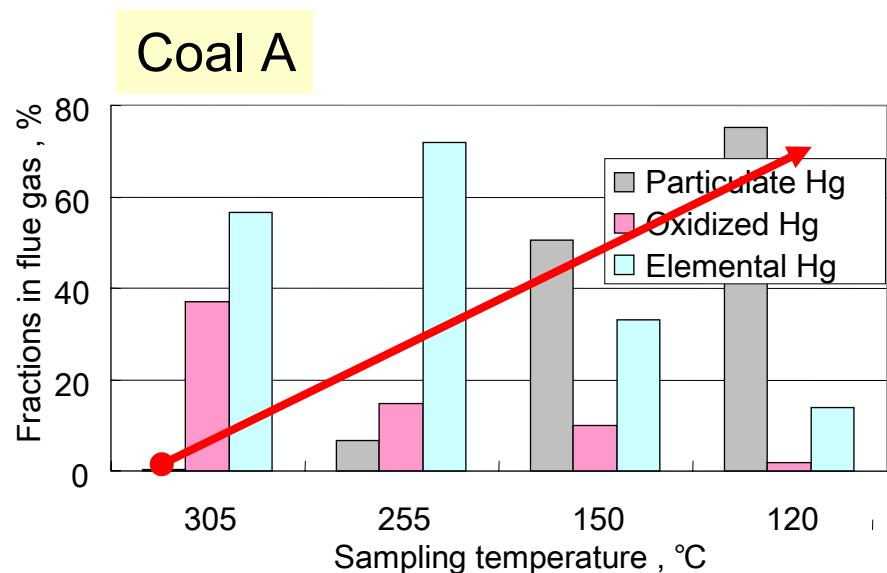


2) Mercury speciation in flue gas at low temperature



Mercury in dust is increased with decrease in flue gas temperature.

2) Mercury speciation in flue gas for coal type



Coal A
 Lowering temp \rightarrow $Hg(p) \uparrow + Hg^0 \downarrow$
Coal E with high volatile
 Lowering temperature \rightarrow a little
Coal D with high Cl
 Even high temperature $\rightarrow Hg^+$

**Unburned carbon content
Chlorine content**

Condition of sorbent injection

Coal	5.9
Feed rate, kg/hr	
Residence time, s	
P2 (230°C)	0.706
P4 (170°C)	1.25
P6 (130°C)	1.88

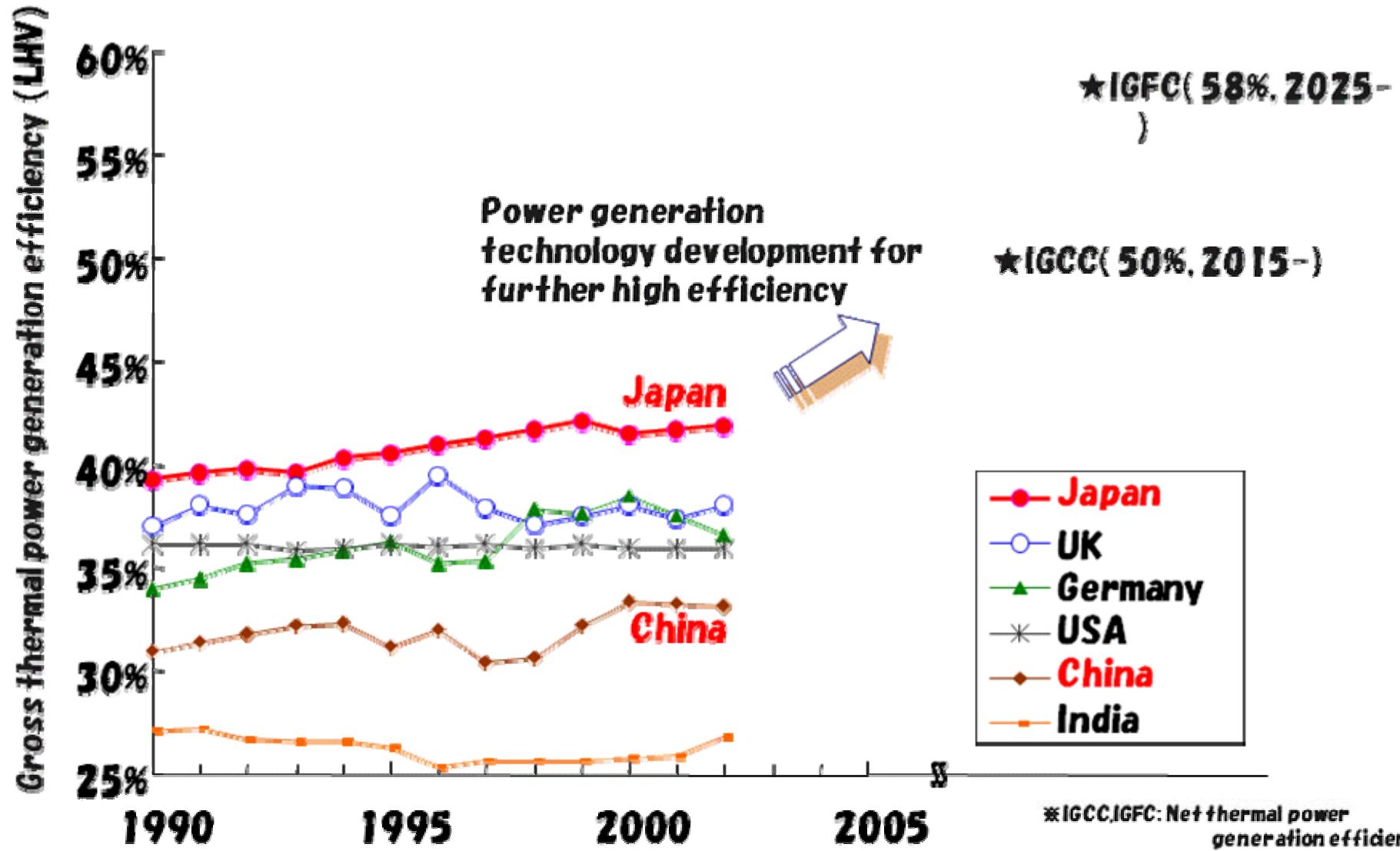
Sorbent feed rate, g/hr	P6 130°C	P4 170°C	P2 230°C
AC from br-coal	42.5	38.9	37.1
Spent catalyst	43.0	38.0	27.8
Coal fly ash	26.9	38.0	23.9

Sor bent			Act ivat ed car bon	Coa l f ly ash	Spent cat alyst
Pr oximate analyses	Moist ure	wt %	12.5	1.4	0.8
	Ash	wt %	4.3	66	98.3
	Vola tile	wt %	6.5	6	0.9
	Fixe d car bon	wt %	76.7	26.6	0
Ult imate analyses	Car bon	wt %	77.2	29.02	0.39
	Hydro gen	wt %	1.91	0.58	0.16
	Nit rogen	wt %	0.22	0.62	0
	Total- Sulphur	wt %	0.2	0.8	0.004
	Mer cur y	wt %	17	3.01	<1
Size		μ m	<125μ m	<100μ m	vary
Dens ity		kg/ m3	ND	758.7	972.05

From cyclone

P2=4930mm P4=8360m P6=11790mm Inside diameter=57.2mm

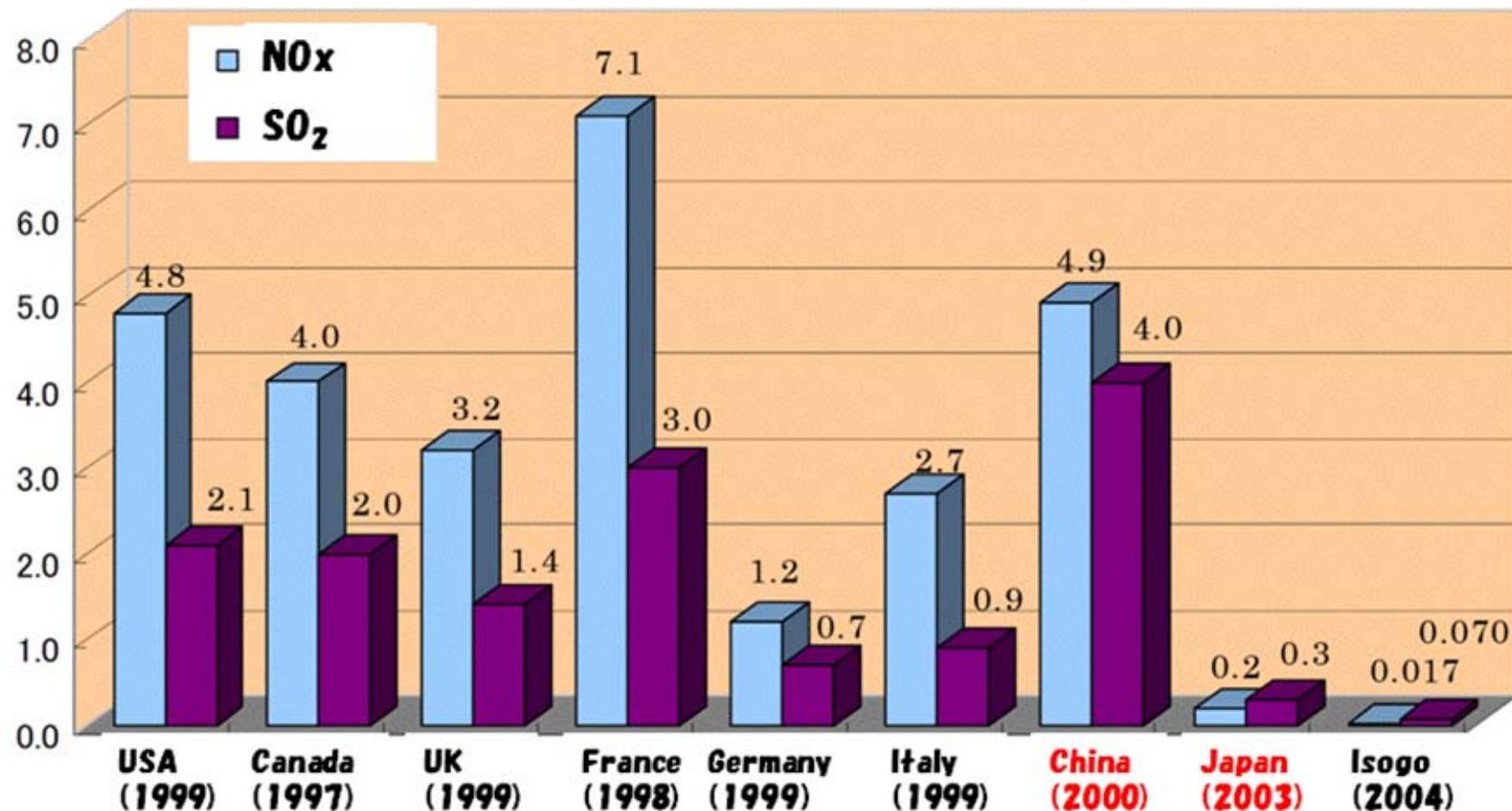
Trends of coal-fired power generation efficiency in each country



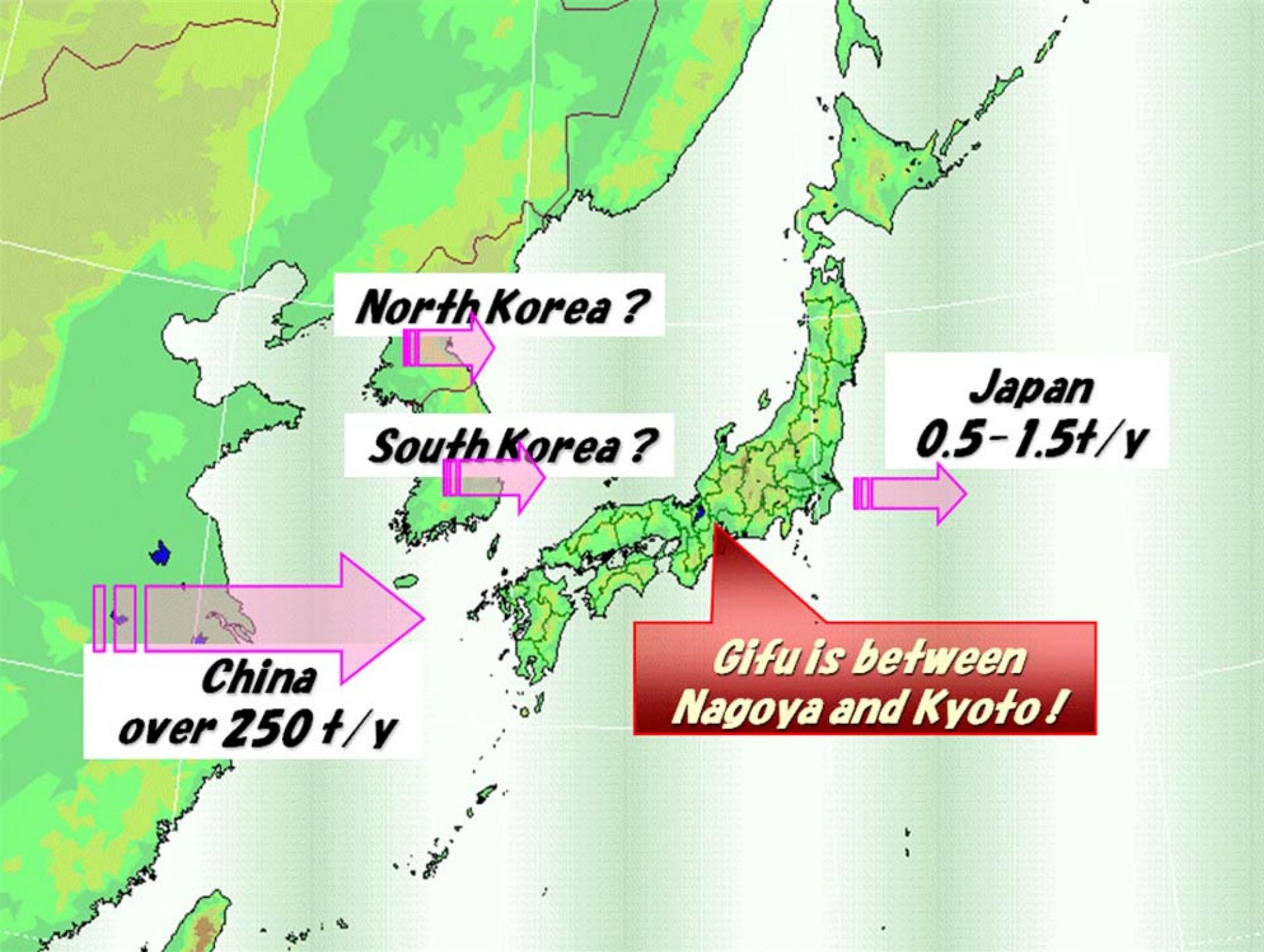
Source: Ecofys Comparison of Power Efficiency on Grid Level 2004

Emission of NOx and SO₂ from each country, g/kWh

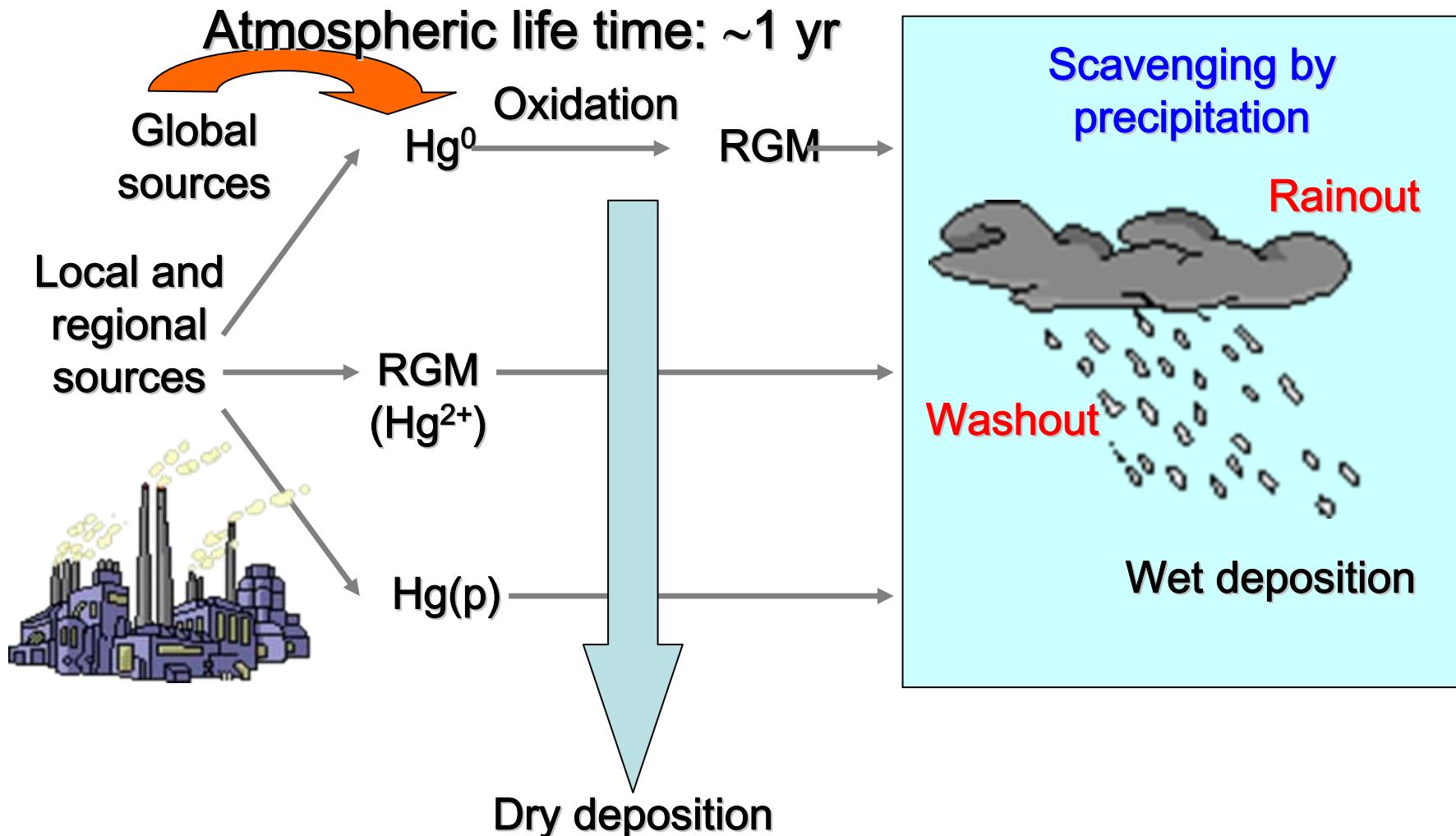
[g/kWh]



Data from METI

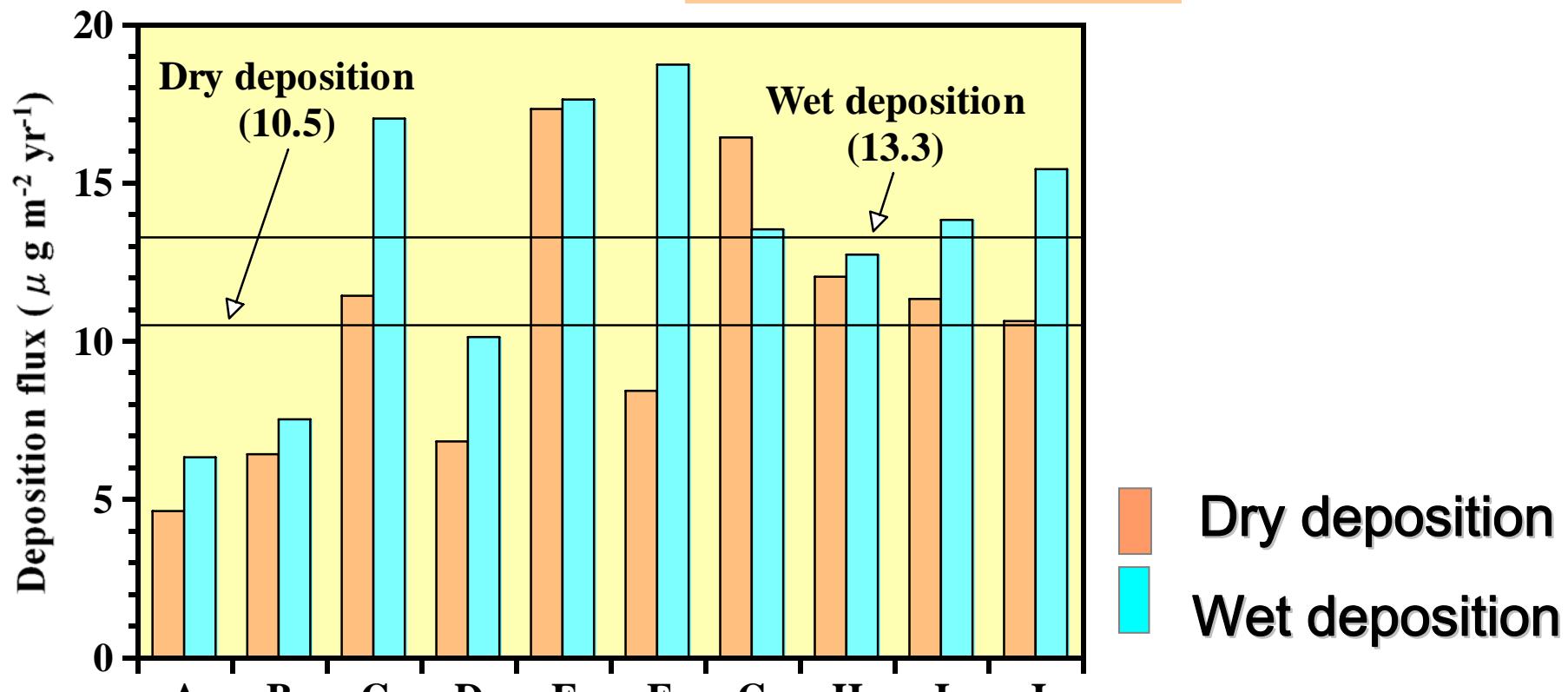


Atmospheric deposition of mercury



Dry and wet deposition fluxes of mercury around power station site

Average for 2003-2005



CRIEPI : Sakata's data

Conclusions

Under the practical conditions of flue gas cooling process of PC boilers, the test results support the following conclusions:

1. Mercury transformation depended on **chlorine** content in coal and **unburned carbon** in flyash dust.
2. Excessive **activated carbon** could capture elemental mercury but a little of oxidized mercury.
3. When using **inorganic sorbents**, there was a **trade-off relationship** between oxidized mercury and elemental mercury and the total captured mercury was lowered.

“Earlier” is “better”
“If slowing, it is not in time.”

Think about what we can do !